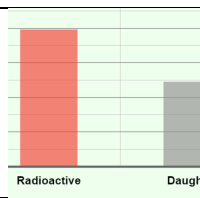
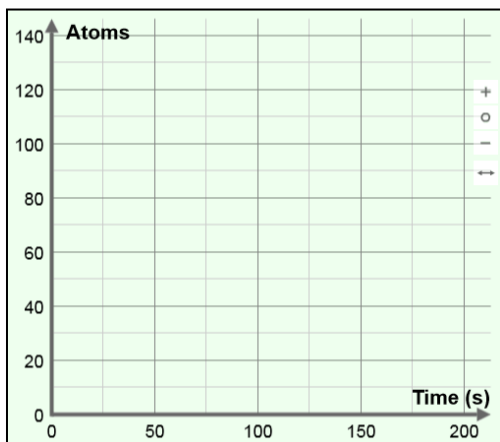


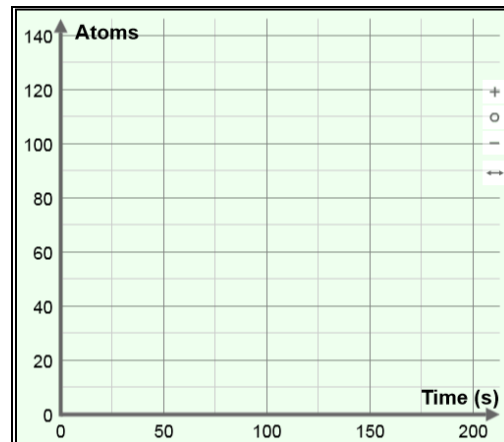
<p><b>Activity A:</b> <b>Decay curves</b></p>	<p>Get the Gizmo ready:</p> <ul style="list-style-type: none"> <li>• Click <b>Reset</b> (↺). Be sure that <b>User chooses half-life</b> and <b>Random decay</b> are selected.</li> <li>• Check that the <b>Half-life</b> is 20 seconds and the <b>Number of atoms</b> is 128.</li> </ul>	
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**Question: How do we measure the rate of radioactive decay?**

1. Observe: Select the BAR CHART on the right side of the Gizmo and click **Play**.
  - A. What happens to the numbers of radioactive and daughter atoms as the simulation proceeds? \_\_\_\_\_  
\_\_\_\_\_
  - B. Do the numbers of radioactive and daughter atoms change at the same rate throughout the simulation? Explain. \_\_\_\_\_  
\_\_\_\_\_
2. Experiment: Click **Reset**, and select the GRAPH tab. Run a simulation with the **Half-life** set to 5 seconds and another simulation with the **Half-life** set to 35 seconds. Sketch each resulting decay curve graph in the spaces below.



**Half-life = 5 seconds**



**Half-life = 35 seconds**

3. Interpret: How does the **Half-life** setting affect how quickly the simulated substance decays?  
\_\_\_\_\_  
\_\_\_\_\_

**(Activity A continued on next page)**



**Activity A (continued from previous page)**

4. Collect data: Click **Reset**. Change the **Half-life** to 10 seconds and click **Play**. Select the **TABLE** tab and record the number of radioactive atoms at each given time below.

0 s: \_\_\_\_\_ 10 s: \_\_\_\_\_ 20 s: \_\_\_\_\_ 30 s: \_\_\_\_\_ 40 s: \_\_\_\_\_ 50 s: \_\_\_\_\_

5. Analyze: What pattern, if any, do you see in your data? \_\_\_\_\_

\_\_\_\_\_

6. Revise and repeat: Use your data from #4 above to fill in the first line of the data table below. Then repeat the experiment four more times. Calculate the average number of radioactive atoms for each time.

Trial	0 s	10 s	20 s	30 s	40 s	50 s
1						
2						
3						
4						
5						
<b>Averages:</b>						

7. Analyze: A **half-life** is defined as the amount of time it takes for half of the radioactive particles to decay. For the simulated substance, every 10 seconds represents one half-life.

How does your data demonstrate the definition of a half-life? \_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

8. Revise and repeat: Click **Reset**. Real radioactive samples will contain billions of radioactive atoms. To model the decay of a large sample, change from **Random decay** to **Theoretical decay** on the **SIMULATION** pane. Click **Play** and record the numbers of radioactive atoms:

0 s: \_\_\_\_\_ 10 s: \_\_\_\_\_ 20 s: \_\_\_\_\_ 30 s: \_\_\_\_\_ 40 s: \_\_\_\_\_ 50 s: \_\_\_\_\_

How does this data demonstrate the meaning of half-life? \_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

